

Black Carbon and Aerosol Analysis in Rain and Snow During SUPRECIP - 2

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Introduction:

Transport of Asian BC (Black Carbon) across the Pacific Ocean is significant during March and April (*Hadley et al. submitted June 2006*)

Transport occurs predominantly at higher elevations (>1 km) (*Heald et al. 2006*) and therefore may impact mountain snowpack albedo and melt rates (*Jacobson MZ, 2004; Hansen J and Nazarenko L, 2004;*)

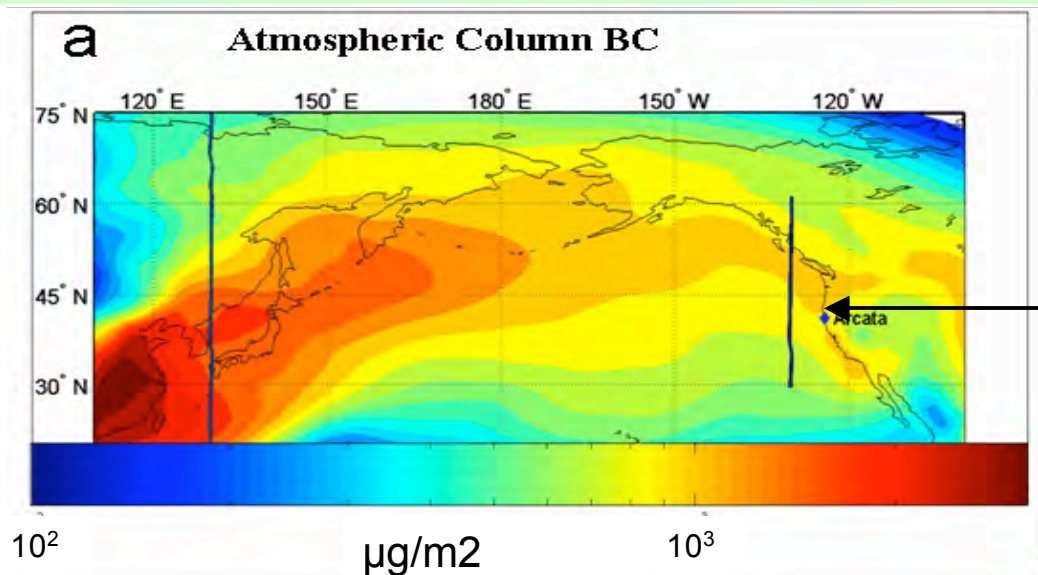
We aim to measure the concentration of BC in falling snow and rain to:

(a) compare BC in coastal rainwater to inland mountain snow fall and

(b) determine if it is large enough to impact snowpack lifetime in the California mountains.

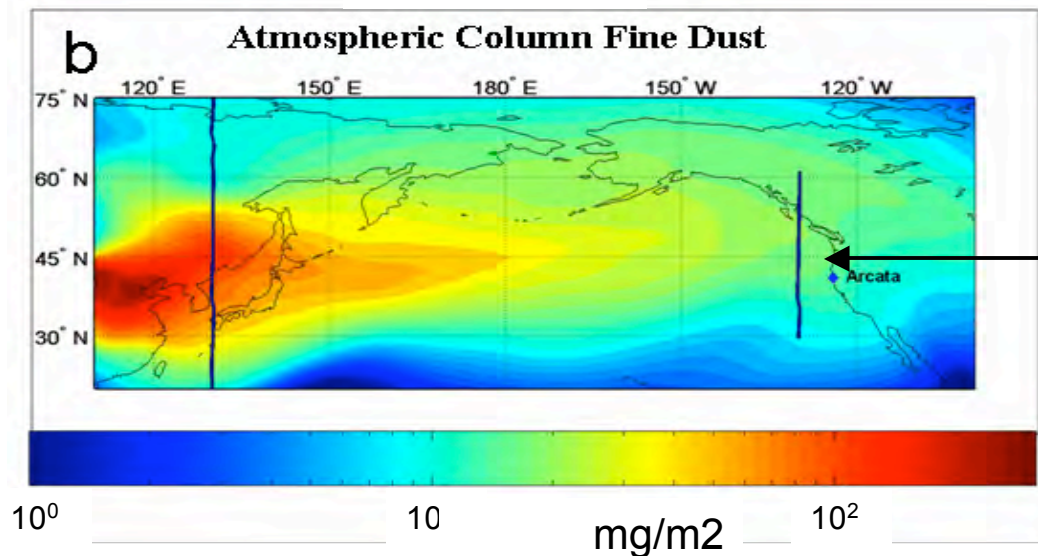
Motivations: Long range transport of aerosols

CFORS (Chemical FORecast System model - April 2004



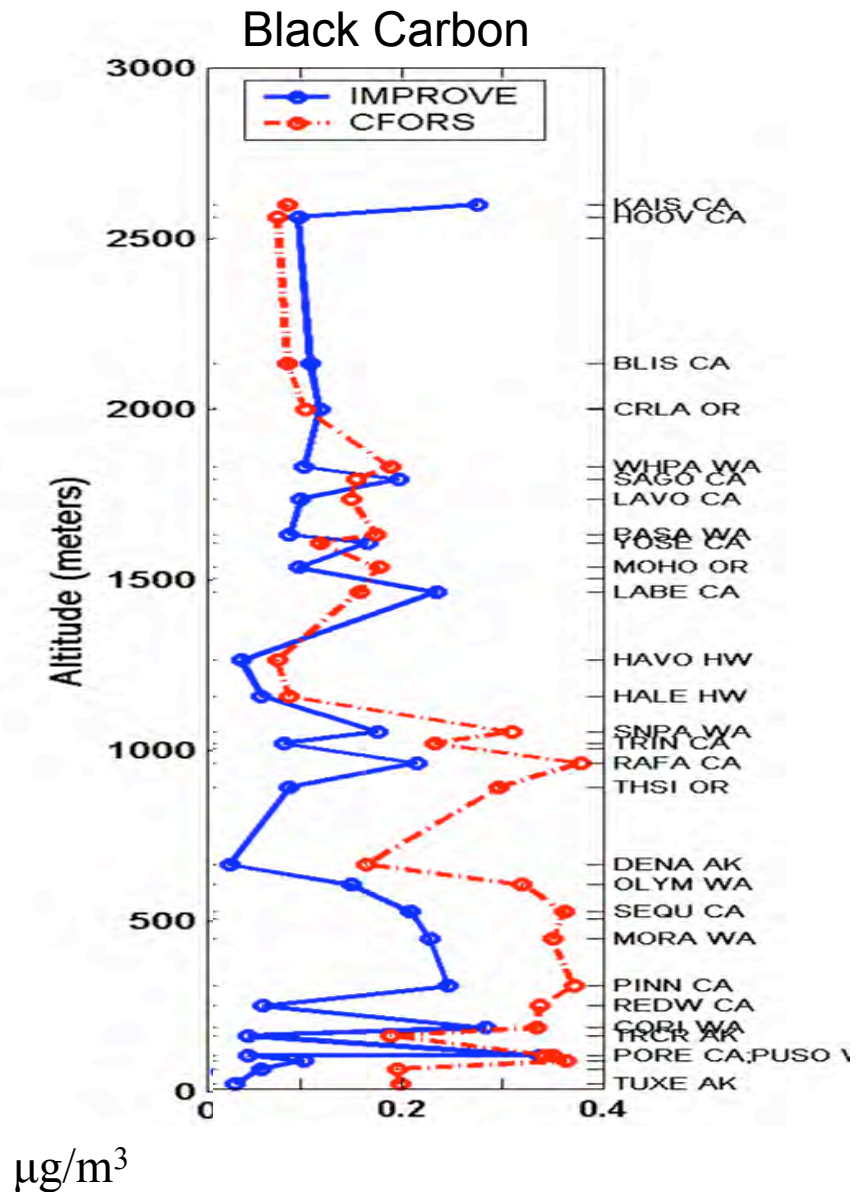
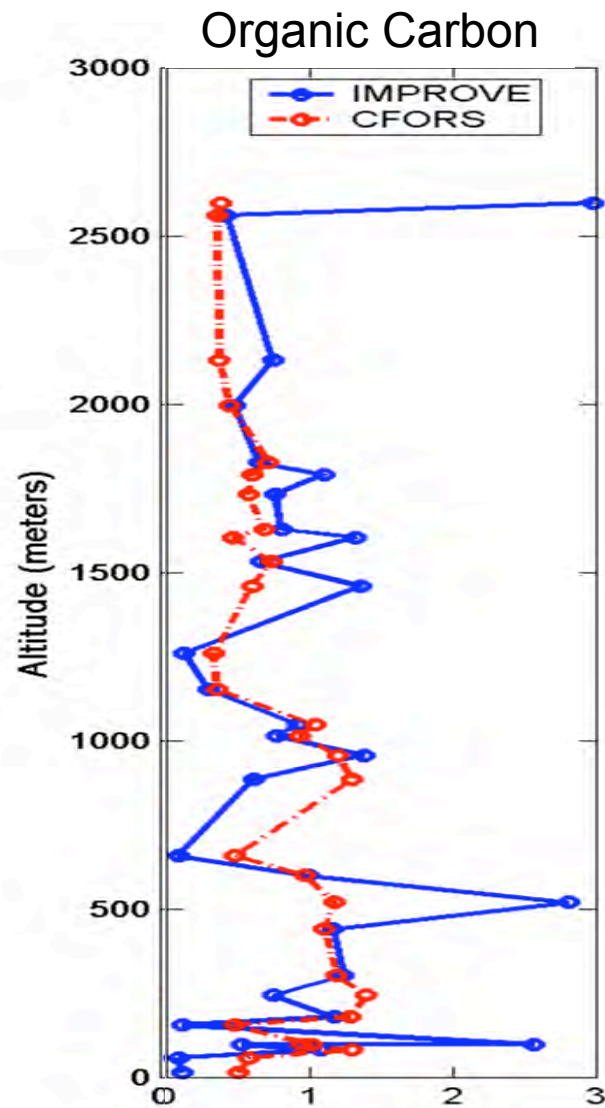
BC total transport
25 – 32 Ggrams

77% of estimated North American monthly BC emissions (*Hadley et al. subm. 2006*)



Fine Mass transport
900 – 1100 Ggrams

Predictions (CFORS) vs. observations



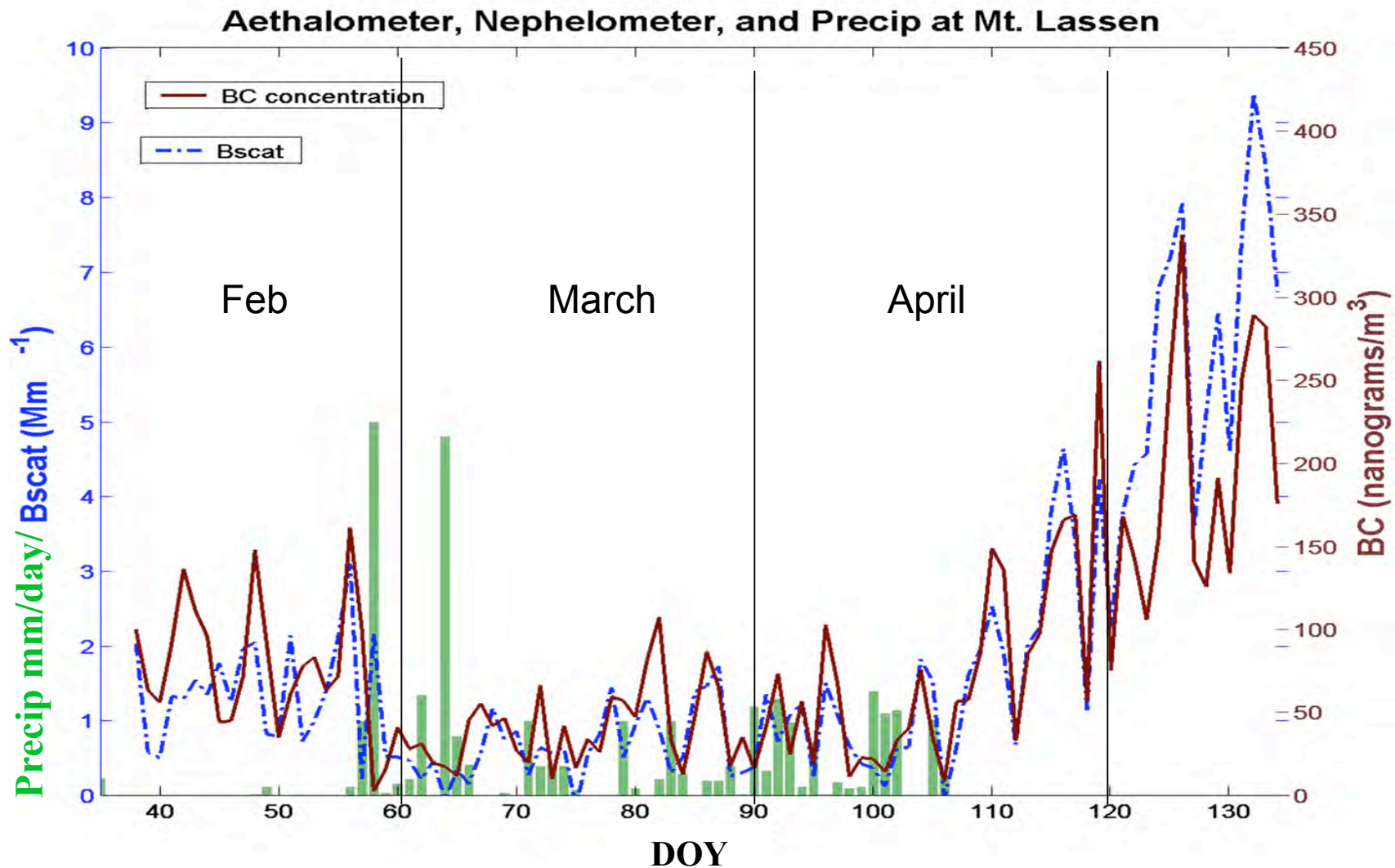
$\mu\text{g}/\text{m}^3$

Goals of the experiment

- **Measure black carbon concentration in snow and rain, both upwind and downwind of local pollution sources.**
 - **Are BC concentrations significant enough to enhance springtime melting of the mountain snow pack?**
- **Look at chemical components in the precipitation as an indicator of origin.**
 - **Could Asian aerosols be influencing snow melt?**



Ambient BC and light scatter at Mt. Lassen Natl. Park 2006



Successful Snow Collection Season



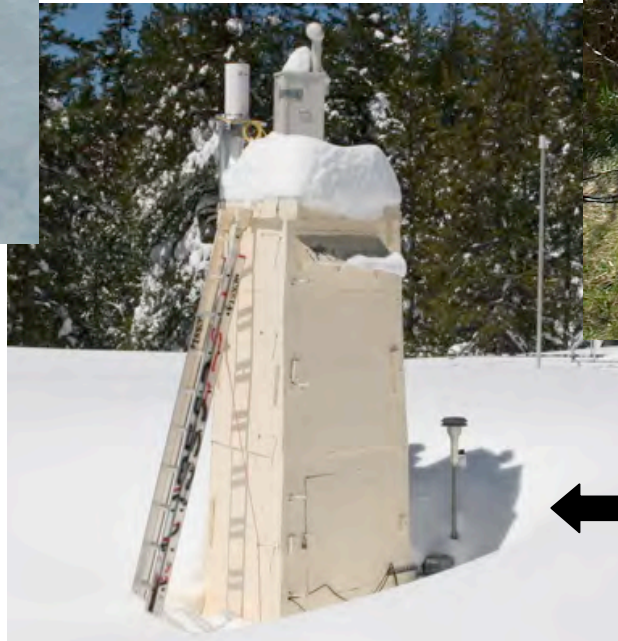
- Site in operation from Feb 26 to April 30th.
- Over 20 samples collected at CSSL





↑
 Lassen Volcano NP
 02/06/2006 – 04/30/2006

Trinidad Head, Arcata
 02/13/2006 – 04/30/2006

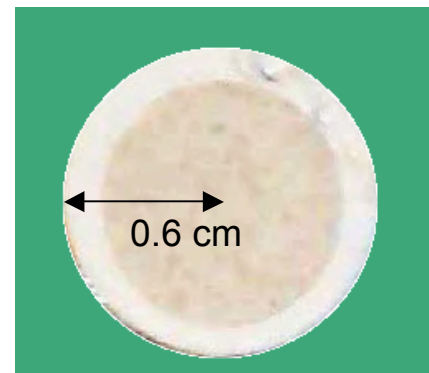
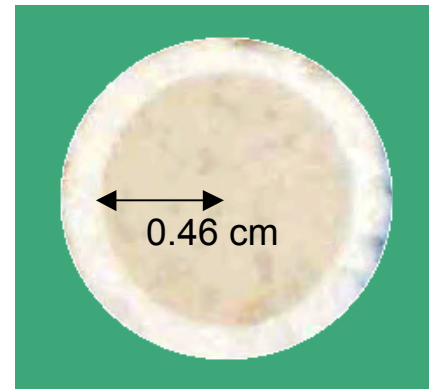


← Central Sierra Snow Lab -
 02/26/2006 – 04/30/2006

Over 80 samples collected at all 3 sites combined.

Black Carbon observed in Precipitation

- 200 mL of precipitation sample are vacuum filtered through quartz fiber filters.
- Color change indicates not only BC, but the presence of yellow or red dust in the rain sample.



Accurate Measurements of Black Carbon are very Difficult to Make!!!

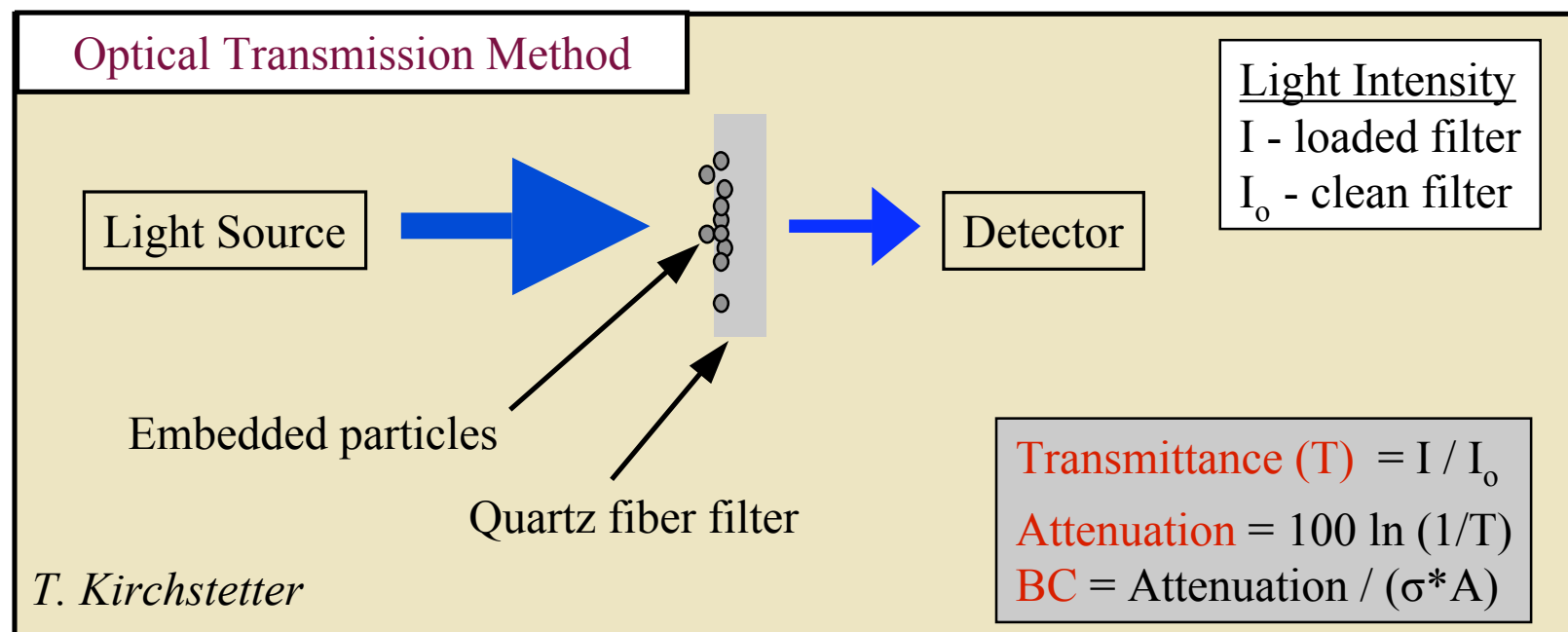
- We use 2 independent methods to determine BC concentration.
- Method 1: Evolved Gas Analysis
- Method 2: Spectral Light Transmission Method.

Method 1. Evolved Gas Analysis, Thermo-Optical Transmittance (EGA-TOT)

- **Filters are heated from 50°C to 800°C in an oxygenated atmosphere**
- **Organics are oxidized to CO₂ at T < ≈ 500°C and BC or soot at T > ≈ 500°C**

$$BC = \int_{T \approx 500^{\circ}C}^{T=800^{\circ}C} \frac{\left(P_{CO_2} \left(\frac{\text{moles } CO_2}{\text{mole } O_2} \right) * O_2 \left(\frac{\text{moles}}{L} \right) * 0.2 \left(\frac{L_{O_2}}{\text{min}} \right) * 12 \left(\frac{g}{\text{mole } C} \right) \right)}{\left(40 \left(\frac{^{\circ}C}{\text{min}} \right) * 1.15 \right)} dT$$

Method 2. SLTM

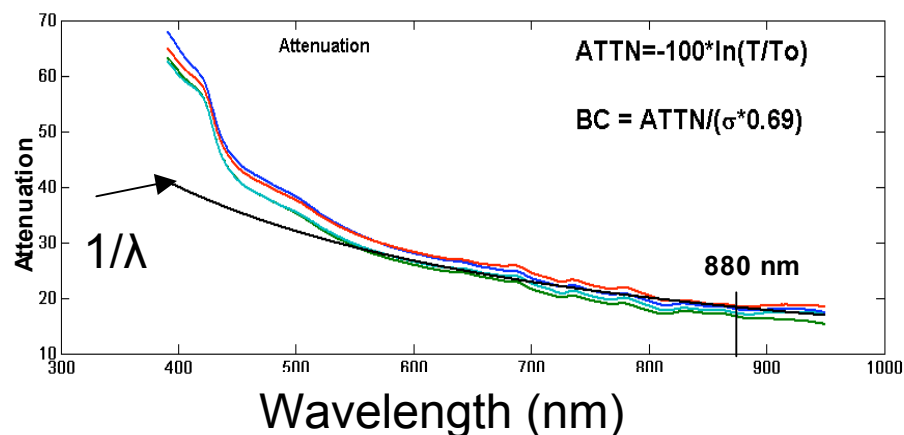
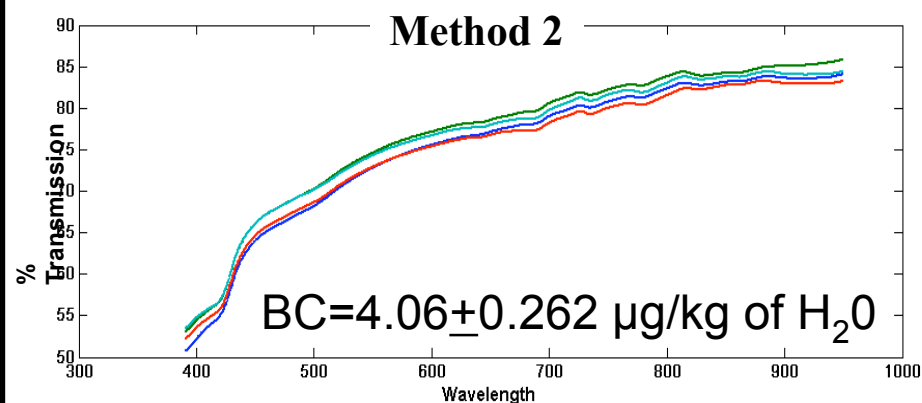
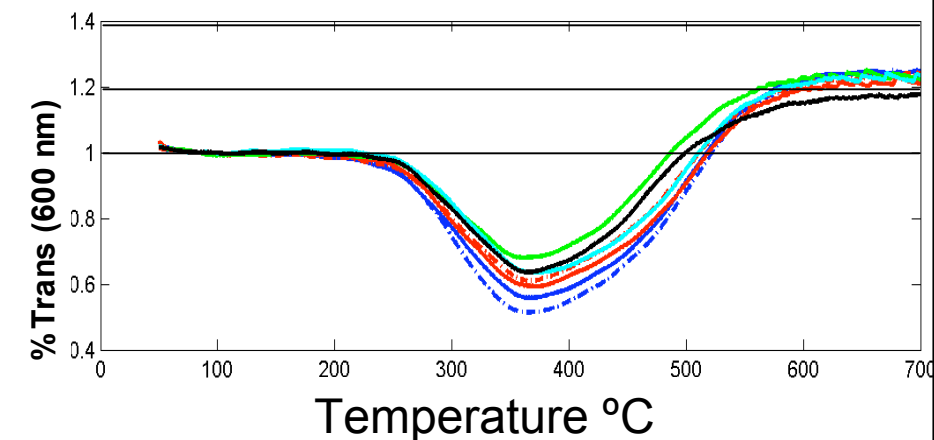
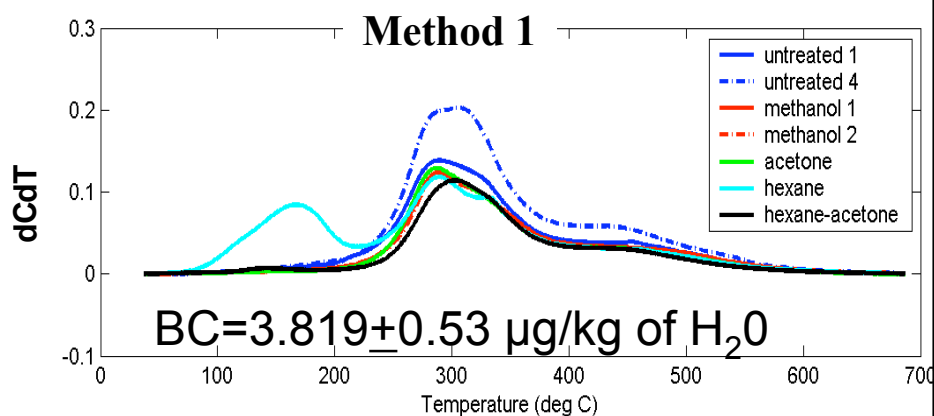


- Method is sensitive mainly to particle light absorption, not scattering
- Widely used to measure aerosol light absorption: Aethalometer, Particle Soot Absorption Photometer (PSAP)
- Continuous light spectra from (370-1200 nm), BC mass is taken at 880 nm
- Sigma (mass absorption efficiency) is set at $20 \text{ m}^2 \text{ g}^{-1}$; $A = 0.69 \text{ m}^2$

EGA-TOT vs. SLTM

Tested 7 filters made from Trinidad Head bulk gauge water bucket.

- Standard deviation is 5% for SLTM and 13% for EGA-TOT.
- Agreement between the two methods is within 6%



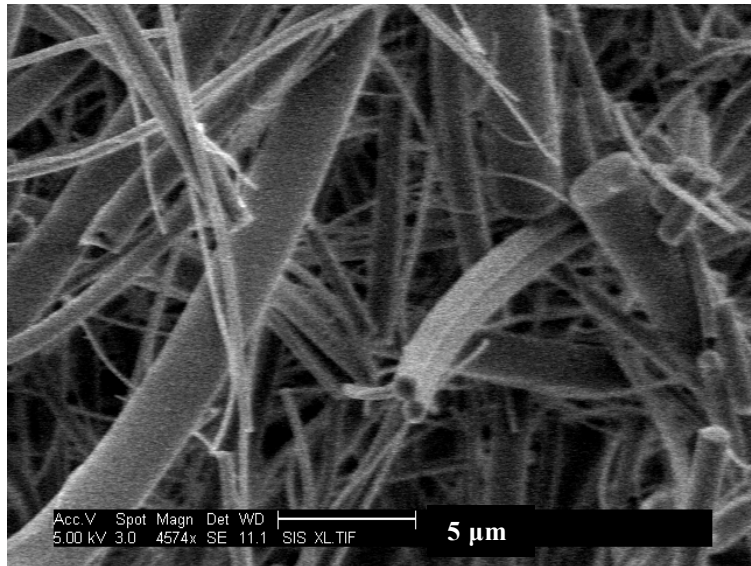
Challenges with measuring BC in Water

- **Determine efficiency of BC collection on the quartz fiber filter (these filters must be used for both types of analysis)**
 1. **Use SEM to look at sizes of particles that pass through the filter.**
 2. **Evaluate the efficiency of the filter using a known BC standard.**
- **Evaluate BC losses to surfaces during collection and analysis**

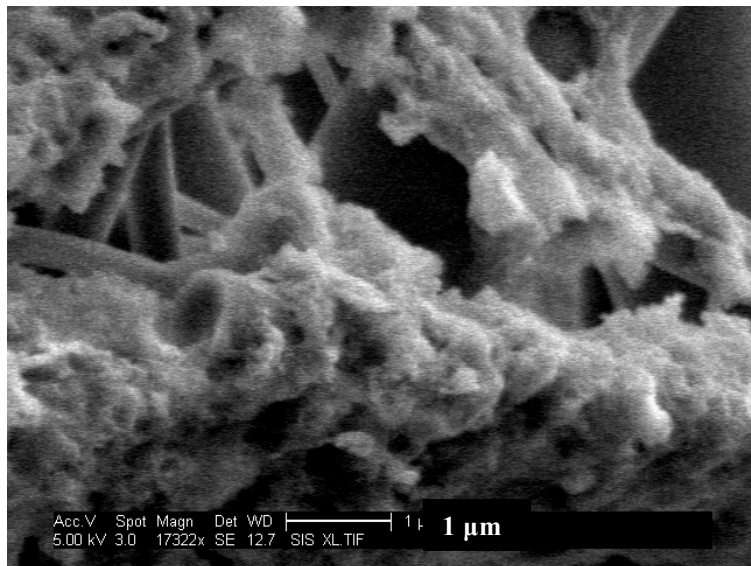
SEM images of Snow Melt from Lassen

Quartz Fiber

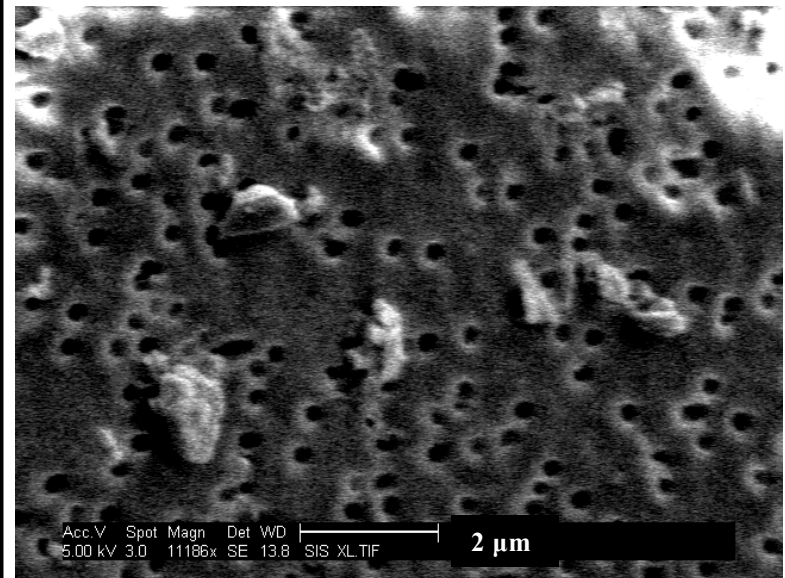
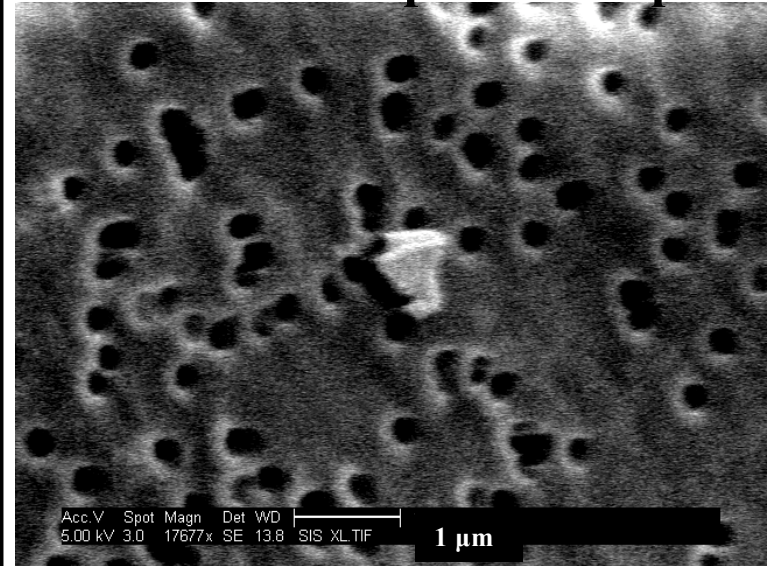
followed by → 0.45 micron pore Nuclepore



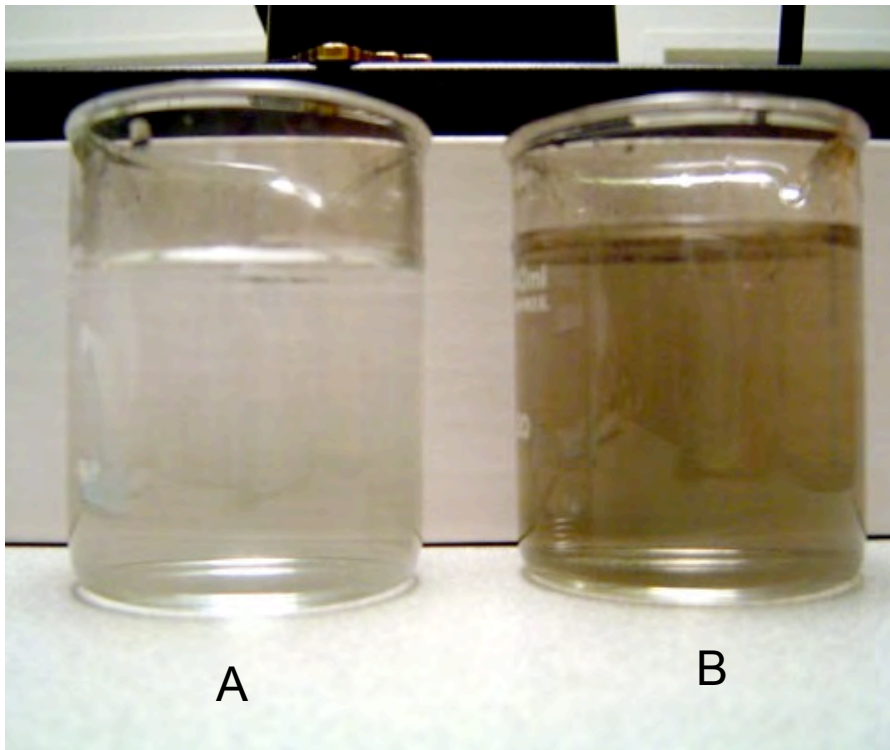
← Clean



← Dirty



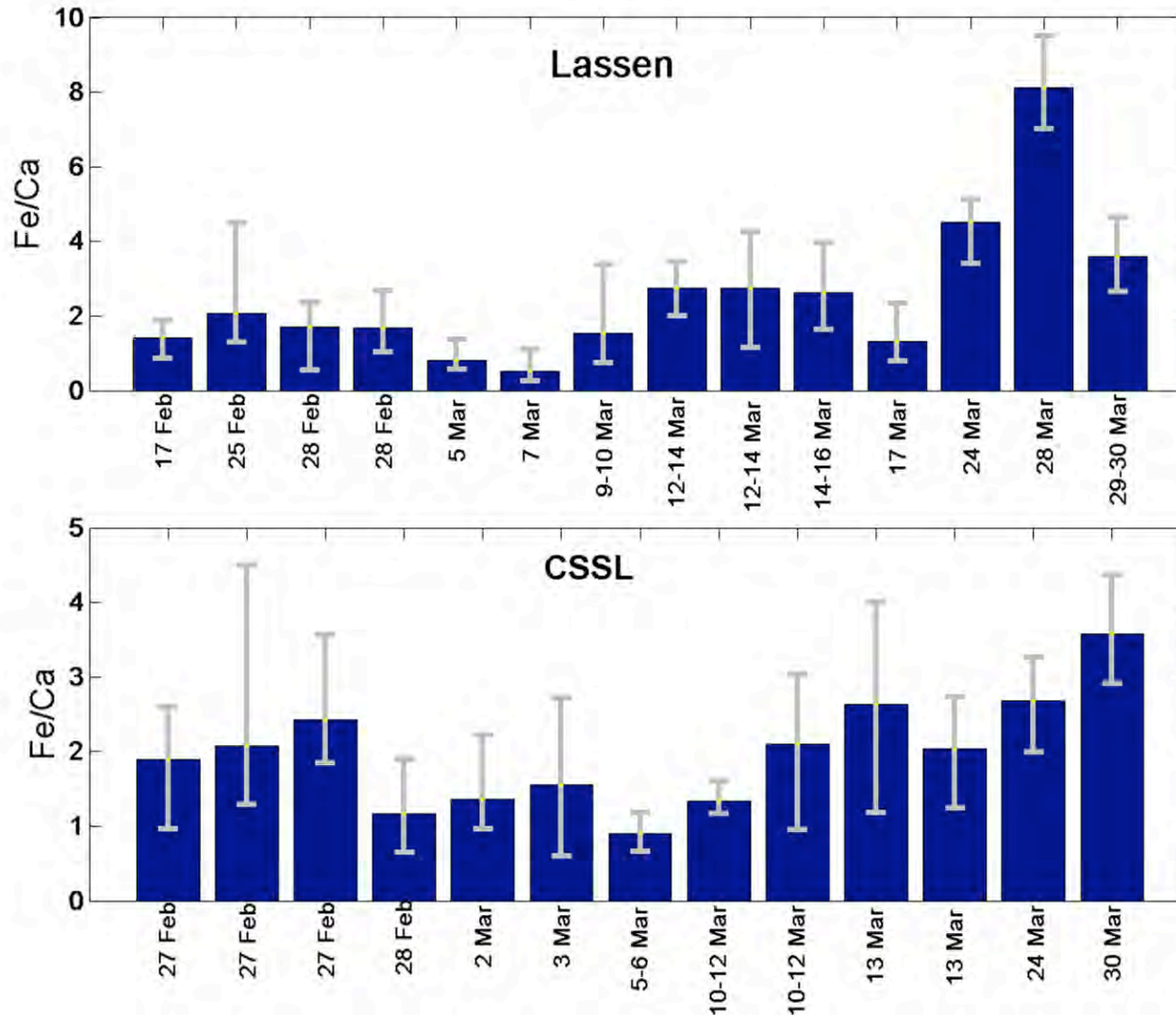
Standard for BC in water to calibrate filter efficiency



T. Kirchstetter's group 2006

- Standard is used to calibrate filter efficiency.
- To get a suspension of soot in water it must be artificially aged by reacting with ozone.
- The soot is then bubbled through distilled water to make the standard.
- A) no ozone
- B) ozone reacted

Mt. Lassen Volcano Natl. Park and Central Sierra Snow Lab: XRF analysis of snow melt water



Further work...

- **Complete measurements of BC concentration in rainwater and snow meltwater.**
- **Compare the evolution of BC and elemental signatures as a storm moves from the coast to Sierras.**
- **Place BC concentrations into context with previous work on solar absorption by soot and enhanced snowpack melt.**

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